

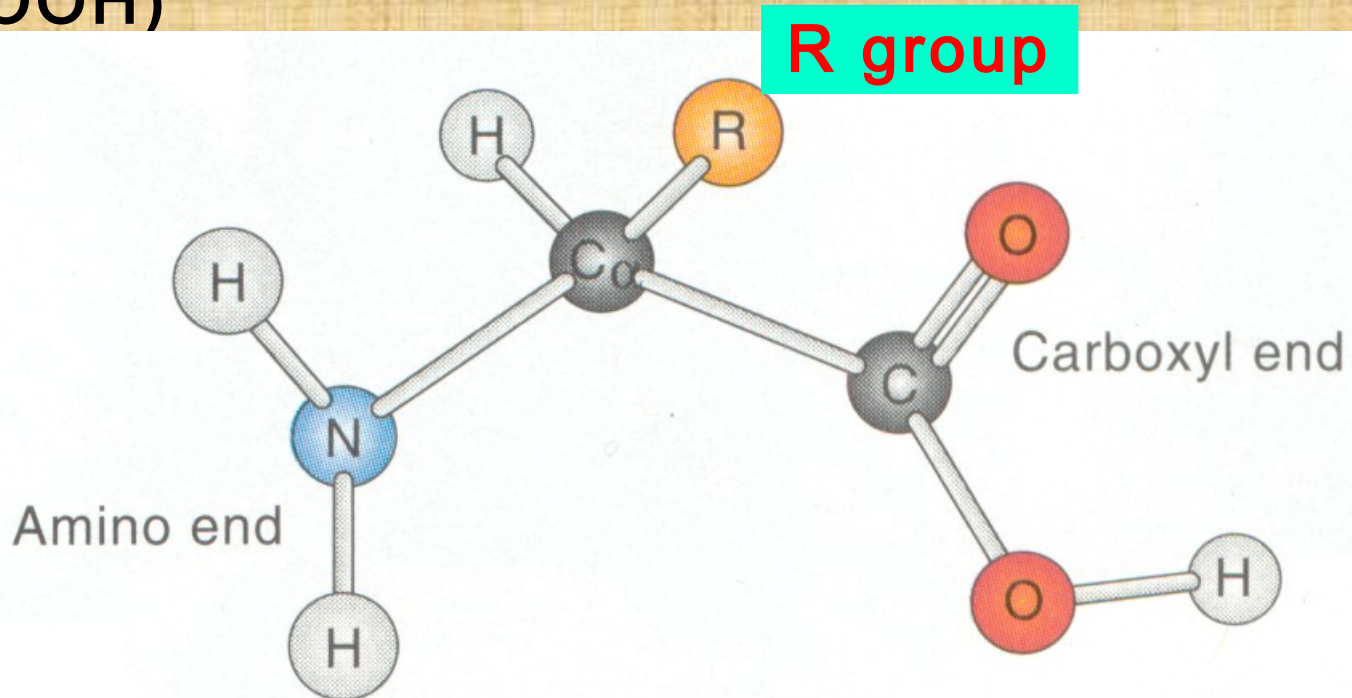
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# Amino Acids

# What is amino acid?

## What is amino acid?

- Amino Acid: aminated carboxylic acid (R-COOH)



# Classification of Amino Acid

1. By the location of Amino-group :  $\alpha$  /  $\beta$  /  $\gamma$ -AA
2. By its acidity : **neutral/ acidic/ basic** AA  
ratio of Amino-group to carboxylic group
3. By whether containing phenyl group  
**aromatic / non aromatic** AA
4. By its occurrence in protein  
**Protein / non protein** AA
5. By polarity of R group :  
**polar / apolar** side chain AA
6. By its nutrient value to human:  
**Essential** AA and **non-essential** AA



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# PROTEIN ANALYSIS

# What is Protein ?

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- **polymer** of 20  $\alpha$ - amino acids, with mol.wt from 5000 to 1000,000 daltons.
- N is most **distinguished** element: among the composing elements of C, H, N, O, S, for some proteins: P, Cu, Fe, I.
- N content in different proteins ranging from 13.4% - 19.1%, and **averagely 16%**.
- **Most abundant component in cells: 50% of dry cells by weight**

# Protein content in food in %

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**Cereals :** ( % )

Brown Rice 7.9

Polished rice 7.1

Wheat flour, whole-grain 13.7

Corn flour, whole-grain 6.9

Corn starch 0.3

# Protein content in food

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## legumes :

Soybean, raw	36.5
Beans, kidney, raw	23.6
Tofu, raw, regular	8.1



# Protein content in food

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## Fruits & vegetables :

Apple, raw, with skin	0.2
Strawberry, raw	0.6
lettuce , raw	1.0



# Protein content in food

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## Dairy products :

Milk, whole, fluid	3.3
Milk, skim, dry	36.2
Cheese, cheddar	24.9
Yogurt	5.3

# Protein content in food

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## Meat, poultry, fish:

Beef	18.5
Dry beef	29.1
Chicken, breast meat, raw	23.1
Ham	17.6
Egg, raw, whole	12.5
Finfish, raw,	17.9

# Conversion factors for Foods

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## N to Protein conversion factors

<b>Foods</b>	<b>factors</b>
<b>Egg or meat</b>	<b>6.25</b>
<b>Dairy products</b>	<b>6.38</b>
<b>Wheat</b>	<b>5.70</b>
<b>Other cereal grains and oilseeds</b>	<b>6.25</b>
<b>Almonds</b>	<b>5.18</b>
<b>Peanuts</b>	<b>5.46</b>
<b>Other tree nuts and coconut</b>	<b>5.30</b>



# Kjeldahl's method

## Principles:

1. **Digest** the organic compounds with strong sulfuric acid in the presence of catalysts while heating.
2. The **total organic N** is converted to **ammonium sulphate**.
3. **Neutralize** the digested sol'n with abundant alkali. Here, the N is converted to **ammonium hydroxide**, and then being distilled into a boric acid solution and converted to **ammonium borate**.
4. **Titrate** ammonium borate with strong acid.  
(please notice that N: HCl = 1:1)
5. N content in proteins is averagely 16%.

# Equipments

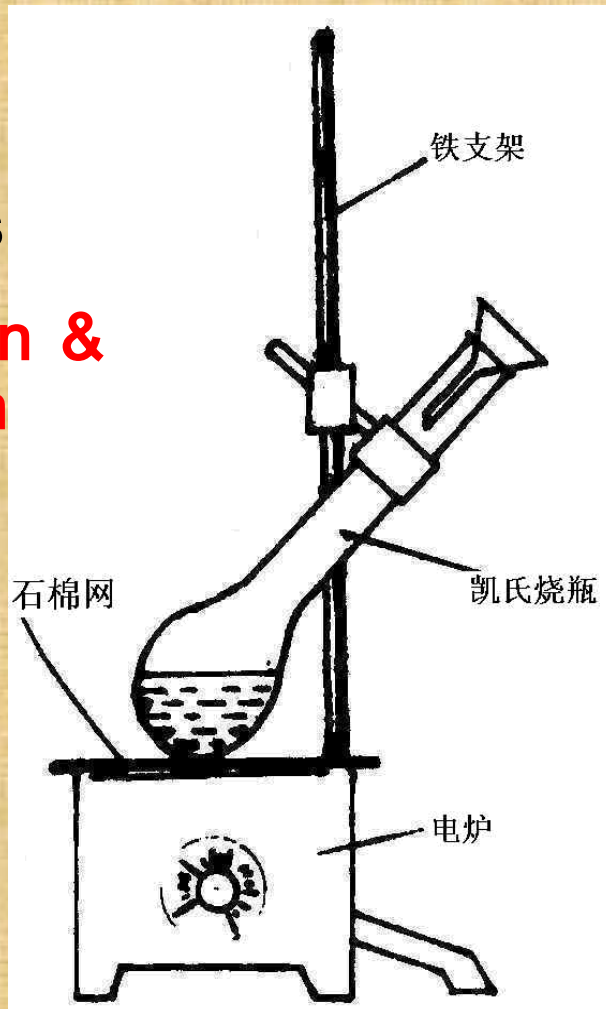
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- -a. Kjeldahl digestion flask - 500 or 800 ml
- b. Kjeldahl distillation apparatus,
- c. Conical flask, 250 ml
- d. Burette 50 ml.

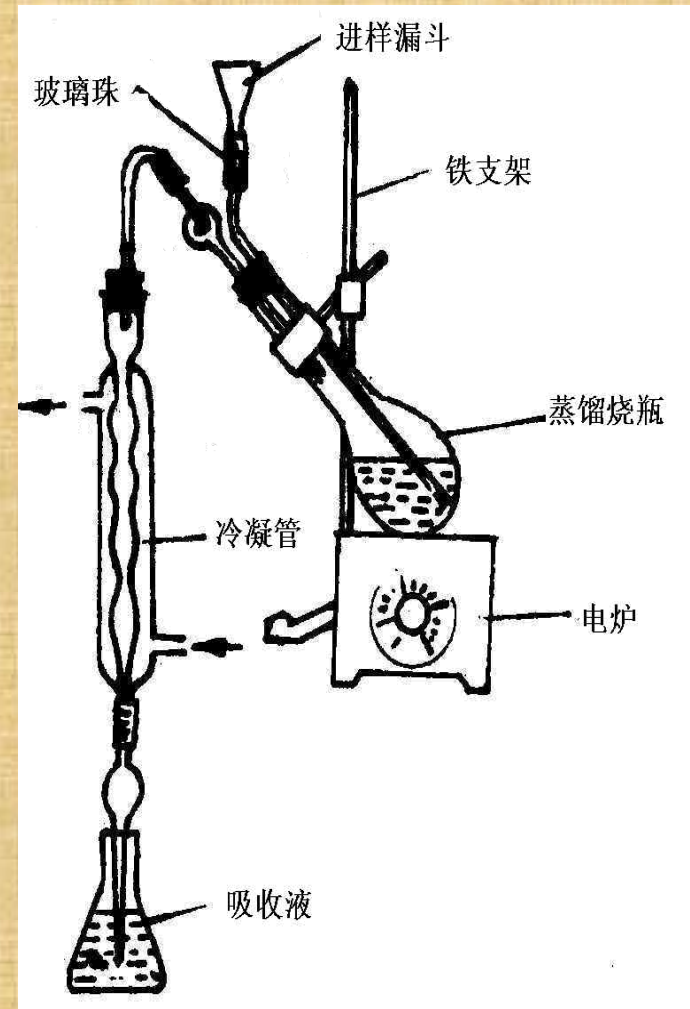
# Apparatus used in Kjeldahl

I. Digestion apparatus

II. Distillation & absorption apparatus



(I)



(II)



# Procedure

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- Add 2gm of sample+0.7gm mercuric oxide+15gm potasium sulfate
- Add zinc granules
- Heat mixture gently for 90mnts until the froathing ceaces
- Then boiled directly and continued digestion
- Cool the contents

# Procedure

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- Add 150ml of water to the cooled digestion
- Transfer the content to R.B flask
- Connect to kjeldhal assembly
- Add 40% NaOH solution carefully through the sides
- Add zinc granules
- Distill all ammonia to boric acid
- Titrate with 0.1N NaOH solution (indicator-methyl red+methylene blue)

# Calculations

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- % Nitrogen (as is) =  
(ml 0.1N H<sub>2</sub>SO<sub>4</sub> - Blank - ml 0.1N NaOH) × 0.0014 × 100 , the whole divided with sample wt.

$$\% \text{ Protein} = \% \text{ Nitrogen} \times 6.25$$



# Points that need your close attention

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1. Amount of protein sample and reagents used should be **proportional**.
2. All the working solution should be prepared with **ammonia-free distilled water**
3. **Mildly heating** When digestion, so that no sample to spatter onto flask wall.
4. **Rotate the flask** while digestion.
5. **Add antifoam** (silica oil) if necessary.

# Points that need your close attention

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6. Connect well the distillation apparatus **before** adding alkali into digested solution.
7. Cold water bath is a good choice to lower the temp.
8. Using **indicating paper** to help for the determination of distillation terminus.
9. **Indicators** of methylene blue and methyl red are added to absorption bottle before carrying in the distillation

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THANKS